

## **Science Virtual Learning**

# MPI Physics 240 Thermodynamics 21: Cyclic Processes May 20, 2020



#### Lesson: MPI Thermodynamics 20 Cyclic Processes May 20, 2020

Objective: To understand thermodynamic processes that go through a repetitive cycle of steps, like an engine This video discusses systems that go through a repeating set of thermodynamic steps, and how they are represented on p-V diagrams

https://youtu.be/C7tJlcYoySg

Video: Cyclic Processes



An engine is comprised of a piston, which contains a diatomic gas. The gas starts with a volume of 1.00 L and pressure of 1.00 atm at 300 K (Point A). Heat is then added to the piston at constant volume until the temperature reaches 900 K (Point B). The piston is allowed to expand isothermally back to a pressure of 1.00 atm (Point C). The piston is then compressed isobarically back to its starting point (Point A).

a) Draw a p-V diagram of the process

b) Calculate p, V, and T at points A, B, and C.

c) Calculate Q and W for each step of the cycle (A $\rightarrow$ B, B $\rightarrow$ C, C $\rightarrow$ A).

d) Calculate the efficiency of this engine.

Video: <a href="https://youtu.be/e3GCRbbBQOE">https://youtu.be/e3GCRbbBQOE</a>

### Example Video

#### Homework

An engine is comprised of a piston, which contains a diatomic gas. The gas starts with a volume of 8.00 L and pressure of 1.00 atm at 300 K (Point A). The piston is then compressed isothermally until it reaches a volume of 1.50 L (Point B). The gas is then allowed to expand isobarically back to its original volume of 8.00 L (Point C). Heat is then removed from the gas at a constant volume until it reaches its starting point (Point A).

a) Draw a p-V diagram of the process

b) Calculate p, V, and T at points A, B, and C.

c) Calculate Q and W for each step of the cycle ( $A \rightarrow B, B \rightarrow C, C \rightarrow A$ ).

d) Calculate the efficiency of this engine.

- Try to solve the problem yourself, then watch the solution video:
- https://youtu.be/62pLS74HxSw

### That's it!